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Modern Chemistry Chapter 5-Final Review. STUDY. PLAY. Periodic law. The physical and chemical properties of the elements are periodic functions of their atomic numbers. periodic table. An arrangement of the elements in order of their atomic numbers so that elements with similar properties fall in the same column, or group.

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Describe.... 1) Dalton's model - solid sphere, made up of only 1 type of matter, no space. 2) Schrodinger's model - based on probability of finding an e-; electrons are found in orbitals (look like electron clouds), which are in sublevels (s,p,d,f,g), which are in energy levels.

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lanthanide. actinide. alkali metals. The
physical and chemical properties of the
elements are perio.... The 14 elements
with the atomic numbers 58-71. The 14
elements with the atomic numbers
90-103. The elements of group 1 of the
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CHAPTER 5 REVIEW The Periodic Law
SECTION 1 SHORT ANSWER Answer the following questions in the space provided.

1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered.

5 The Periodic Law

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