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Skill Practice 23

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3 0.30 0.90 0.30 0.40 0.40 0.80 3.350 x 10⁻⁴ M/s 9.045 x 10⁻³ M/s 5.600 x 10⁻³ M/s a) What is the order with respect to HI in this reaction? b) What is the order with respect to CH₃I in this reaction? c) What is the overall order in this reaction? d) Write the rate law for this reaction. e) Calculate the rate constant and include units.

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Skill Practice 19 - S.W.H.S CHEMISTRY

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Skill Practice 1

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Skill Practice 13 Name: _____ Date: _____ Hour: _____ 1. An atom has a mass number of 43 and it also has 21 electrons. a) How many protons does this atom have? b) What is the identity of this atom? c) How many neutrons does this atom have? 2. What is an isotope? Give an example. 3.

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Skill Practice 30 Conversion Practice Name: _____ Date: _____ Hour: _____ 1. Calculate the number of molecules in 210 grams of 'water. 7.02x10² molecules 2. If you have 6.25 x 10² molecules of sulfur tetrafluoride, how many grams do you have? 1120 g 3. Define what a mole is. Don't give a number, give a definition in words.

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3. Consider the wave in the previous question. Why don't you know the amplitude of the wave? 4. Consider a wave that has a wavelength of 3.7 m. If the frequency was 2.5 Hz, then calculate the speed of the wave. 5. A wave travels at a speed of 12.5 m/s. The frequency is 24.3 Hz. What is the wavelength?

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3. How many liters of NH₃ were produced? How many grams of NH₃ is this? The pressure in the lab is 97.8 kPa and the temperature was 23.7°C. 2. If 22.5 L of oxygen reacted with excess hydrogen, how many liters of water vapor (gaseous water) could be produced? 3. The combustion of a certain wax can be represented by the following balanced ...

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Skill Practice 11 Name: _____ Date: _____ Hour: _____ 1. What is wrong with the following notation? 2. How many sublevels would you expect in the 8th energy level? 3. What is the maximum number of electrons that can fit in the 3d sublevel? 4. How many electrons can fit in a 2p orbital? 5.

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