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Practice Problems In your notebook, answer the following SECTION 7.1 IONS For each element below, state (i) the number of valence electrons in the atom, (ii) the electron dot structure, and (iii) the chemical symbol(s) for the most stable ion. a. Ba 2.

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The molecular structure of carbon  
dioxide is one carbon atom with two

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oxygen atoms on opposite sides of it.

\_\_\_\_\_ 13. Covalent bonds exist when combining atoms give up or accept electrons. \_\_\_\_\_ 14. A molecule contains two atoms. Part C Matching Match each description in Column B to the correct term in Column A.

**05 CTR ch08 7/12/04 8:12 AM Page**

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**181 MOLECULAR COMPOUNDS 8**

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Class IONIC AND METALLIC BONDING

Practice Problems In your notebook,  
answer the following SECTION 71 IONS

For each element below, state (i) the  
number of valence electrons in the  
atom, (ii) the electron dot structure, and  
(iii) the chemical symbol(s) for the most

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unhealthy and that, perhaps, the grass is dying.

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